

Resolution, Spectra, Dipole Strength, and Rotational Strength of (+)-(1*S*,5*S*)-Bicyclo[3.2.0]heptan-3-one

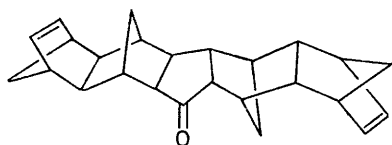
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Summary Enantiomerically pure (+)-(1*S*,5*S*)-bicyclo[3.2.0]heptan-3-one has been prepared and characterized, and its c.d. spectra, rotational strength, and dipole strength have been determined; its rotational strength is unusually large.

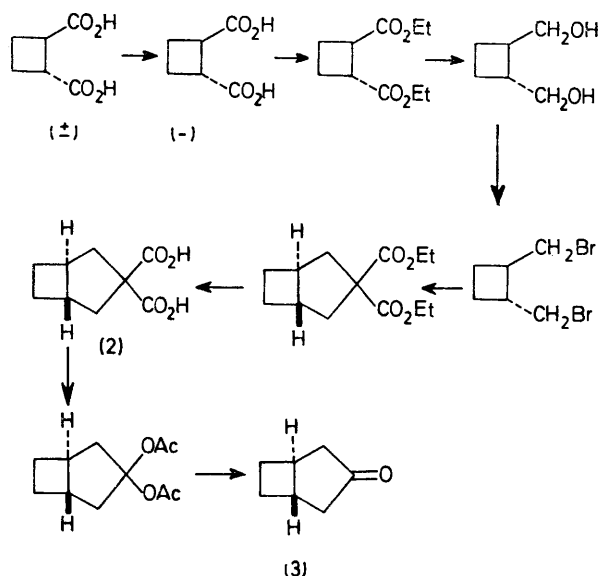
RECENTLY Weissberger¹ reported the exceptional rotatory power of the dissymmetric cyclopentanone (**1**), but since its u.v. and c.d. spectra were recorded in different solvents

values for its rotational strength (R.S.) and dipole strength (D.S.) cannot be compared. We have prepared (**3**), another dissymmetric cyclopentanone, from (-)-(1*R*,2*R*)-cyclobutane-1,2-dicarboxylic acid,^{2,3} according to the procedure for racemic (**3**),^{4,5} *via* (-)-(1*S*,5*S*)-bicyclo[3.2.0]heptane-3,3-dicarboxylic acid (**2**) (Scheme). Crude (**3**) was purified by preparative g.l.c. (Varian A 90, OV 210, *T*_c 110 °C)³ and characterized by i.r., u.v., c.d. ¹H n.m.r. and mass spectroscopy, and comparison with data for



(1)

racemic (3).⁴ The absolute configuration of (3) was established by applying the octant rule,^{3,6} V.F.F.-R.S. calculations,^{3,7} and by comparison with homologous



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dissymmetric bicyclic cyclopentanones,³ all three methods giving the same result. The enantiomeric purity of (3), determined by circular polarization of luminescence^{3,8} was ca. 100%. U.v. and c.d. spectra were recorded in iso-octane at room temperature with a spectral resolution of 1 nm (Figure).

† The experimental values for R.S. and D.S. were obtained from the u.v. and c.d. spectra where ϵ and $\Delta\epsilon$ were plotted as functions of $\ln \bar{\nu}$, and the area under the curve determined; cf. also S. F. Mason, *Quart. Rev.*, 1963, 17, 20.

‡ The dissymmetry factor is small enough [$g_{\max}(\lambda) = 0.22$] to neglect deviations from the Bouguer-Lambert-Beer law; cf. also F. Woldbye and S. Bagger, *Acta Chem. Scand.*, 1966, 20, 1145.

Pure (3) showed: $[\alpha]_D + 568^\circ$, $[\phi]_D + 625^\circ$, (c 0.333, iso-octane); u.v. λ_{\max} (iso-octane): 290, 300, 311, and 323 nm (ϵ : 41.51, 52.65, 48.29, and 24.70, respectively); c.d. λ_{\max} (iso-octane): 290, 300, 311, and 323 nm ($\Delta\epsilon$: 6.87, 9.73, 9.87, and 5.31, respectively); ^1H n.m.r. $\delta[(\text{CD}_3)_2\text{SO}]$ 0.49–2.10 (m); i.r. (neat) ν_{\max} 1743s cm^{-1} (C=O); m/e 110; D.S. 6.5×10^{-51} (S.I. units); † R.S. = 22.0×10^{-53}

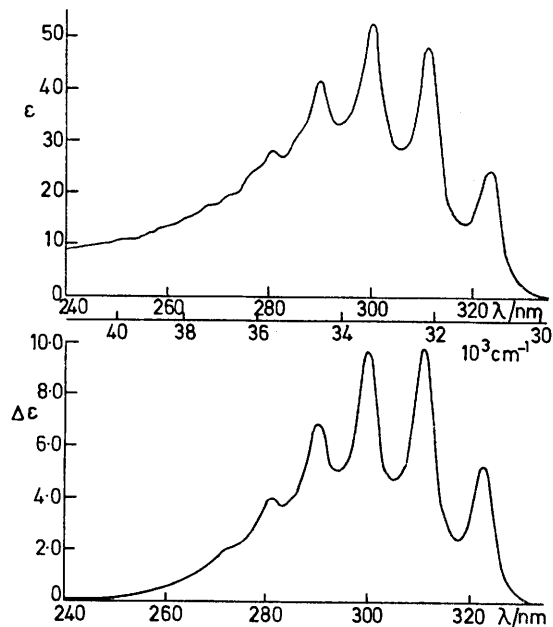


FIGURE C.d. spectra of (3).

(S.I. units). †‡ Although Weissberger does not give a value for the R.S. of (1) we estimate it to be ca. $6\text{--}7 \times 10^{-52}$ S.I. units from its c.d. spectrum. The R.S. value for (3) is unusually large for a saturated ketone; the absolute value for the R.S. of $(-)$ -(1*R*,6*R*)-bicyclo[4.3.0]nonan-8-one is 17.0×10^{-53} S.I. units.⁴

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⁴ J. Meinwald, J. J. Tufariello, and J. J. Hurst, *J. Org. Chem.*, 1964, 29, 2914.

⁵ J. J. Tufariello and W. J. Kissel, *Tetrahedron Letters*, 1966, 6145.

⁶ W. Moffitt, R. B. Woodward, A. Moscovitz, W. Klyne, and C. Djerassi, *J. Amer. Chem. Soc.*, 1961, 83, 4013.

⁷ We used valence force field calculations to compute the geometry of (3), and this computed geometry was used in the R.S. calculation (cf. also: ref. 3; and C. Altona and H. Faber, *Fortschr. Chem. Forsch.*, 1974, 1, 45 and references cited therein).

⁸ W. C. M. C. Kokke, *J. Amer. Chem. Soc.*, 1974, 96, 2627.